Additions and Corrections

1972, Volume 11

John A. Ferguson, Thomas J. Meyer*, and David G. Whitten*: Metalloporphyrin Redox Chemistry. Unusual Electron-Transfer Behavior in the Oxidation of Lead Porphyrins.

Page 2770. In column 1, lines 12 and 13, the *g* values from epr measurements should read 2.0158 and 2.0161, respectively.–T. J. Meyer

1973, Volume 12

Frank J. Weigert and John D. Roberts*: Nuclear Magnetic Resonance Spectroscopy. Spin–Spin Coupling of Carbon to Phosphorus, Mercury, Nitrogen, and Other Elements.

Page 315. We are indebted to Professor W. T. Ford for pointing out that the data listed for tri-*n*-propylboron in Table III are incorrect. The broad line corresponding to the α carbon is actually at 31.7 ppm, the β carbon resonance comes at 18.4 ppm, and one from the γ carbon at 17.8 ppm.–J. D. Roberts

R. J. West and S. F. Lincoln*: Exchange of Acetonitrile on Complexes of Nickel(II) and Cobalt(II) Formed with 2,2′,2″-Triaminotriethylamine and 2,2′,2″-Tri(*N*,*N*-dimethylamino)triethylamine.

Page 495. Equation 3 should read:

$$T_{2\text{p}}^{-1} = \left(\frac{\tau_\text{M}}{P_\text{M}}\left[\frac{(T_{2\text{M}}^{-1} + \tau_\text{M}^{-1})^2 + \Delta\omega_\text{M}^2}{T_{2\text{M}}^{-2} + (T_{2\text{M}}\tau_\text{M})^{-1} + \Delta\omega_\text{M}^2}\right]\right)^{-1} + \left(\frac{T_{2\text{M}}'}{P_\text{M}'}\right)^{-1}$$

Page 496. The left-hand ordinate of Figure 2 should read: ln ($T_{2\text{p}}m$).–S. F. Lincoln

G. M. Bancroft,* H. C. Clark, R. G. Kidd, A. T. Rake, and H. G. Spinney: Manganese-55 Nuclear Magnetic Resonance Studies of Compounds Containing Mn–M (M = Si, Ge, Sn, Pb) Bonds and the Signs of e^2qQ in Manganese(I) Compounds.

Page 730. In Table III, the observed e^2qQ for $\text{Mn(CO)}_5\text{SnCl}_3$ should read 7.3 *not* 73.–G. M. Bancroft

J. Pradilla-Sorzano and John P. Fackler, Jr.*: Base Adducts of β-Ketoenolates. V. Crystal and Molecular Structures of *cis*-Bis-(1,1,1,5,5,5-hexafluoro-2,4-pentanedionato)bis(pyridine)zinc(II) and -copper(II).

Pages 1174 and 1175. In the title and several other places the title compounds are incorrectly referred to as *cis*-bis(1,1,1,6,6,6-hexafluoro-2,4-pentanedionato)bis(pyridine)zinc(II) and -copper(II). The correct names are *cis*-bis(1,1,1,5,5,5-hexafluoro-2,4-pentanedionato)bis(pyridine)zinc(II) and -copper(II).–John P. Fackler, Jr.

J. Pradilla-Sorzano and John P. Fackler, Jr.*: Base Adducts of β-Ketoenolates. VI. Single-Crystal Electron Paramagnetic Resonance and Optical Studies of Copper(II)-Doped *cis*-Bis(hexafluoroacetylacetonato)bis(pyridine)zinc(II), $\text{Cu-Zn(F}_6\text{acac)}_2\text{(py)}_2$.

Page 1188. Equation 8 should read:

$$g = \cos^2\left(\frac{\phi_2}{2}\right)g_{x^2-y^2} + \sin^2\left(\frac{\phi_2}{2}\right)g_{3z^2-r^2}$$

R. Morassi, F. Mani, and L. Sacconi*: Spin Isomerism in Five-Coordinate Cobalt(II) Complexes of the Terdentate Ligands *N*-R-*N*-2-(diphenylphosphino)ethyl-*N*′,*N*′-diethylethylenediamine (Donor Set NNP).

Page 1249. Reference 15 should read: R. Morassi and L. Sacconi, *J. Chem. Soc. A*, 1487 (1971).–L. Sacconi

G. B. Robertson and P. O. Whimp*: Crystal and Molecular Structure of a Stable π-Oxopropenylmanganese Tricarbonyl Complex.

Page 1741. In the **Data Collection** section, the formula for the standard deviation of an individual reflection (line 21) should read

$$\sigma(F_\text{o}) = [(LP)^{-2}\Delta I^2 + (\rho|F_\text{o}|^2)^2]^{1/2}/2|F_\text{o}|$$

while on line 34, the formula for $\sigma_\text{s}(F_\text{o})$ should read

$$\sigma_\text{s}(F_\text{o}) = (LP)^{-1}\Delta I/2|F_\text{o}|$$

P. O. Whimp

Earl L. Muetterties: Stereochemical Lability of Eight-Coordinate Complexes.

Page 1964. The second sentence should read "... ; an X-ray analysis of the related $[(n\text{-C}_4\text{H}_9)_4\text{N}]_3\text{Mo(CN)}_8$ salt has shown that the $\text{Mo(CN)}_8{}^{3-}$ ion...".

Page 1966. In Table I, column entries for ^{13}C nmr data were inadvertently transposed. The ^{13}C δ_a and δ_c values for $\text{Zr[S}_2\text{CN(CH}_2\text{CH}_2\text{CH}_3)_2]_4$ are −180.9 and −140.7, respectively. The ^{13}C δ_a and δ_c values for $\text{Ti[S}_2\text{CN(CH}_2\text{CH}_2\text{CH}_3)_2]_4$ are −148.5 and −107.0, respectively.

In the third paragraph the third sentence should read "The electronic spectra of $[(n\text{-C}_4\text{H}_9)_4\text{N}]_4\text{W(CN)}_8$...". The fifth sentence should read "There was, however, no correspondence between the electronic spectrum of the propylammonium salt...".–E. L. Muetterties